## عنوان مقاله:

Thermodynamic modeling of solubility of hydrogen sulfide and carbon dioxide in aqueous diisopropanolamine+ N-(2-aminoethyl) ethanolamine solution

## محل انتشار:

سومین کنفرانس سراسری نوآوری های اخیر در شیمی و مهندسی شیمی (سال: 1395)

تعداد صفحات اصل مقاله: 10

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## خلاصه مقاله:

Natural gas as a clean source of energy contains several contaminates such as CO2 and H2S that is treated through a natural gas purification unit in gas industry. Moreover, for design and construction of gas contactor equipment, it is necessary to obtain experimental values of solubility for H2S and CO2 in aqueous amine/alkanolamines. In this work the Electrolyte-NRTL activity coefficient function is applied to the correlation and prediction of the partial pressure of CO2 and H2S versus the acid gas loading through the absolute average deviation percent, the results show that the present modeling was successful to correlate and predict the binary, ternary, quaternary and five component amine/alkanolamine systems

## کلمات کلیدی:

Acid gas, Electrolyte-NRTL, Thermodynamic modeling